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Note

High-performance liquid chromatographic analysis of *p*-hydroxyacetophenone and *p*-hydroxyacetophenone- β -D-glucopyranoside, two major phenolic compounds in Norway spruce

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p-Hydroxyacetophenone (*p*-HAP) is a major phenolic compound in Norway spruce and plays a significant role in stress phenomena¹⁻³. Under stress conditions, *p*-HAP accumulates in different plant organs. The ratio of *p*-HAP to *p*-hydroxyacetophenone- β -D-glucopyranoside (*p*-HAPG) was shown to increase with the increasing decline-symptoms of Norway spruce. *p*-HAP caused needle-fall, retarded apical growth and inhibited bud-sprouting in biological tests. Therefore, a suitable method for the determination of *p*-HAP and *p*-HAPG was urgently needed to handle a large number of samples in physiological investigations.

We have previously described the measurements of levels of *p*-HAP and *p*-HAPG by a combination of thin-layer chromatography (TLC), enzymic hydrolysis and gas chromatographic (GC)-flame ionization detection (FID)^{1,2}. *p*-HAP and *p*-HAPG were also determined by 2,4-dinitrophenyl hydrazine reagent, TLC and spectrophotometry⁴. The early methods of quantification of *p*-HAPG were by paper chromatography, enzymatically with β -glucosidase^{5,6} or colorimetrically with Milion's reagent^{7,8}.

In the course of applying our previous methods to a study on Norway spruce decline, we became aware of the convenience of high performance liquid chromatographic (HPLC) methods for the determination of biologically active compounds^{1,2}. The present report describes the determinations of *p*-HAP and *p*-HAPG by reversed-phase HPLC, in tissues, needle-segments and whole needles in the nanogram range.

EXPERIMENTAL

Plant materials

Epidermal strips and vascular bundles were detached from needles of Norway spruce trees, *Picea abies* (L.) Korst. (90-100 years old, harvest: June 28th, 1985, FA Bodenmais, Bavaria, F.R.G.) using a microdissection technique. The epidermal strips were cleaned from the mesophyll contaminants by a sharp razor blade. Additionally, needles of cloned Norway spruce (3.5 years old, clone no. 1027, supplied by Pflanzgarten Laufen, harvest: June 19th, 1985) were used.

Chromatographic purification and derivation of authentic samples

Commercially available *p*-HAP (Sigma, Munich, F.R.G.) was purified on a silica gel partition column as described by us¹ to give pure white crystals of *p*-HAP (HPLC purity: 95.87%). Chromatographically purified *p*-HAP was acetylated with pyridine–acetic acid anhydride (1:1) for 4 h at room temperature and purified further by silica gel partition chromatography (column: 35 cm × 1.6 cm I.D., particle size: 230–240 mesh; silica gel + 20% water). Acetyl-*p*-HAP was eluted with *n*-hexane–chloroform (98:2–96:4), in 1% step increases of chloroform, to give pure white acetyl-*p*-HAP crystals (HPLC purity: 94.21%).

The trifluoroacetylation of authentic *p*-HAP (500 mg) was carried out by trifluoroacetic acid anhydride (TFAA)–pyridine (1:1) for 4 h at room temperature and purified by partitioning between chloroform and water (pH 8.0); the purity was checked by high-performance TLC (HPTLC) [($R_{F_{\text{trifluoroacetyl-}p\text{-HAP}}}$ = 0.82–0.87, $R_{F_{p\text{-HAP}}}$ = 0.59–0.61; solvent system: chloroform–ethylacetate (7:3), silica gel F₂₅₄ and HPTLC) (HPLC purity: 95.01%)].

Trifluoroacetylation of p-HAP from plant samples

The trifluoroacetylation of plant samples was performed under the following modified conditions: modification of method 1, Pierce Catalogue (Atlanta, Heidelberg, 1983)–dried sample in 200 μ l of benzene, 100 μ l of 0.01 *M* triethylamine + 100 μ l of TFAA for 10 min at room temperature; addition of 200 μ l of phosphate buffer (pH 6.0), shaken; organic phase (3 ×) collected, evaporated and dissolved in aliquots of ethyl acetate or water–methanol (65:35) prior to injection for GC or HPLC, respectively.

High-performance liquid chromatography–ultraviolet detection

The analysis was done on a Kontron HPLC system (HPLC Pump T-414, UV detector Uvikon 722 LC).

The separation of *p*-HAP and *p*-HAPG was done on a Kontron Spherisorb ODS-2 column (250 mm × 5 mm I.D., 5 μ m particle size) connected with a Vydac 201SC precolumn (50 × 5 mm I.D., 30–40 μ m particle size). The data were acquired and processed by Basic-programmable Shimadzu Integrator Chromatopack C-R3A. The chromatographic conditions are given in the figure legends.

Due to unavailability of radioactive *p*-HAP and *p*-HAP- β -D-glucopyranoside and radioactivity counting equipment in our laboratory, we made some compromises in experimental design. We used acetyl-*p*-HAP as internal standard in our analysis.

Preparation of plant samples for measurement by HPLC

The plant samples were finely homogenized in water and extracted with 80% methanol [+ 100 ppm butylated hydroxytoluene (BHT) as antioxidant]. The extracts were filtered, dried *in vacuo* and diluted in the ratio 1:8000–1:24000.

The internal standard acetyl-*p*-HAP (equivalent to 2.19 μ g) was added in the intermediate dilution of the sample (1:1000) in order to monitor the losses during chromatographic analysis. No influence of BHT could be detected in the analysis under the conditions described below.

Gas chromatography–electron-capture detection

The determination of *p*-HAP and *p*-HAP- β -D-glucopyranoside (after enzymic hydrolysis) following trifluoroacetylation by GC with electron-capture detection (ECD) (^{63}Ni , 10 mCi) was performed under the following conditions: column, 10% SE-30 packed on Gas Chrom Q, 100–120 mesh (1 m \times 2 mm I.D.); column temperature, 130°C; injector temperature, 250°C; flow-rate of nitrogen, 25 ml/min. The measurements were done using a Packard-Becker Gas chromatograph Model 427. Under these conditions, the detection limit of trifluoroacetyl-*p*-HAP was *ca.* 5 ng.

Gas chromatography–flame ionization detection

The preparation of sample and the GC–FID measurement of *p*-HAP and *p*-HAPG were performed according to Hoque².

RESULTS AND DISCUSSION

p-HAP, *p*-HAPG, trifluoroacetyl-*p*-HAP and acetyl-*p*-HAP can be detected at 265 nm (near their maximum absorption wavelength). In our work on the levels of *p*-HAP and *p*-HAPG in Norway spruce needles, we observed no significant interference at this wavelength from the contaminants in the 80% aq. methanolic extracts of Norway spruce needles. Our results are summarized in Tables I, II, III and IV.

Fig. 1 shows the test of linearity of the measurements of *p*-HAP, *p*-HAP- β -

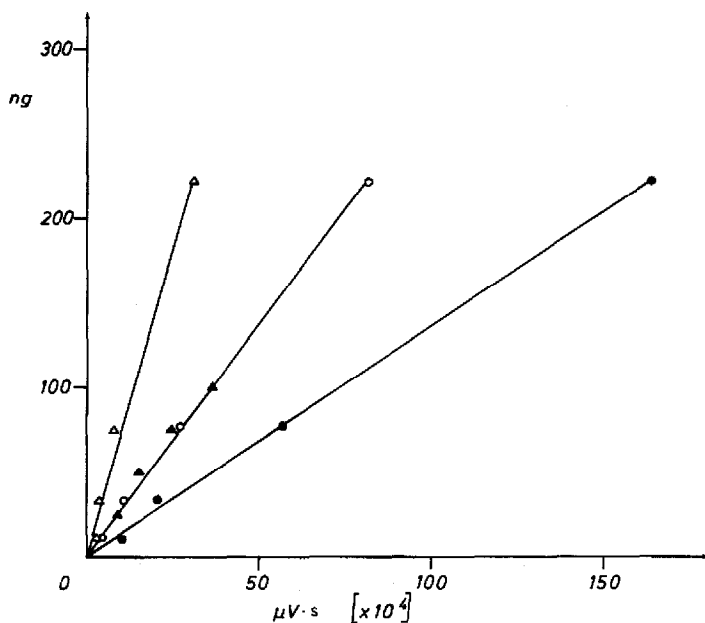


Fig. 1. Test of linearity of the measurement of *p*-HAP (●), *p*-HAP- β -D-glucopyranoside (○), trifluoroacetyl-*p*-HAP (▲) and acetyl-*p*-HAP (△) by isocratic HPLC (0.6 ml/min flow-rate; $\lambda = 265$ nm; mobile phase = 65% water + 35% methanol). The linearity of the measurements of *p*-HAP, *p*-HAP- β -D-glucopyranoside, TFA-*p*-HAP and acetyl-*p*-HAP can be described by the regression lines, $y = 0.51 + 0.000135x$ ($r^2 = 1.00$), $y = 2.39 + 0.000268x$ ($r^2 = 1.00$), $y = 2.46 + 0.000275x$ ($r^2 = 0.99$) and $y = 4.14 + 0.000715x$ ($r^2 = 0.98$), respectively.

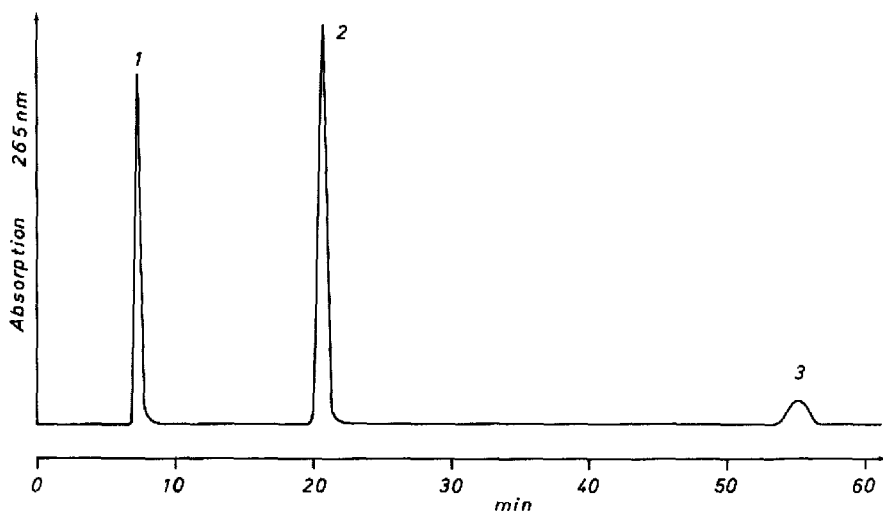


Fig. 2. Separation of authentic samples of *p*-HAPG (1), *p*-HAP (2) and acetyl-*p*-HAP (3) by isocratic reversed-phase HPLC. An aliquot equivalent to 77.77 ng of each sample was injected.

D-glucopyranoside and acetyl-*p*-HAP by HPLC. Fig. 2 shows the separation of authentic *p*-HAP, *p*-HAP- β -*D*-glucopyranoside and acetyl-*p*-HAP, and Fig. 3 shows the separation of these substances from plant material. Phenolic acids could not be separated from each other by the water-methanol (65:35) mobile phase. The elution pattern of the phenolic compounds studied here were in the following order: phenolic

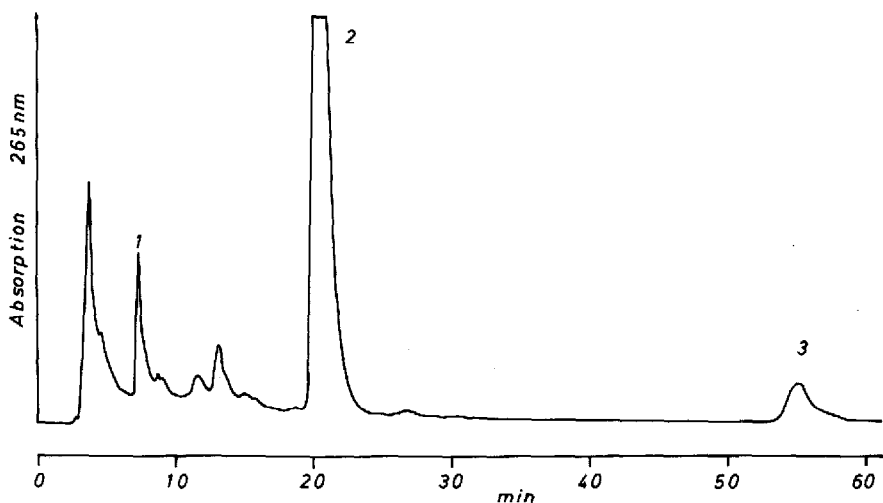


Fig. 3. Analysis of *p*-HAPG (1) and *p*-HAP (2) in plant samples by isocratic reversed-phase HPLC. Peak 3 corresponds to the internal standard acetyl-*p*-HAP. An aliquot of the diluted extract equivalent to 4.17 (affected tree)-250 μ g (healthy tree) fresh weight was injected for measurement. The above chromatogram was obtained from needles of an affected tree (FA Bodenmais, harvest: 28.6.85, ca. 100-year-old tree). The losses of acetyl-*p*-HAP were between 15 and 20% during chromatographic analysis.

TABLE I
RETENTION TIMES OF PHENOLIC COMPOUNDS

Compounds	Retention time (min)
<i>p</i> -HAPG	8.3
<i>p</i> -Cumarinic Acid	5.5
Ferulic Acid	5.5
Caffeic Acid	4.1
Benzoic Acid	5.3
<i>p</i> -HAP	22.5
Trifluoroacetyl- <i>p</i> -HAP	23.0
Acetyl- <i>p</i> -HAP	55.0

TABLE II
PRECISION AND ACCURACY OF THE MEASUREMENT OF *p*-HAP AND *p*-HAPG BY HPLC IN EPIDERMAL STRIPS AND NEEDLES OF NORWAY SPRUCE

p-HAP: $\mu\text{g/g}$ fresh weight, *p*-HAPG: $\mu\text{g/g}$ fresh weight calculated as *p*-HAP equivalents

Object	Tree No.	Substance	Average (\bar{x})	Standard deviation ($n = 10$)	Precision***	Accuracy §
Epidermal strips* (3-year-old needles)	1	<i>p</i> -HAP	329	30	9.0	6.8
		<i>p</i> -HAPG	5781	399	7.0	5.3
Needles** (5 years old)	2	<i>p</i> -HAP	20	3.50	17.0	12.8
		<i>p</i> -HAPG	2531	393	16.0	12.1

* Site Bodenmais (1100 m above sea-level, harvest: June 28th, 1985).

** Site Sauerlach (670 m above sea-level, harvest: June 12th, 1984).

*** Precision = coefficient of variation (C.V.%) = (standard deviation \times 100)/ \bar{x} .

§ Accuracy ($S_{\bar{x}}\%$) = (C.V.% \times t)/ $\sqrt{n-1}$, where t = t value of 2-tail Student's t -test at $P = 0.05$ and degrees of freedom = $n - 1$.

were measured by the TLC-GC method, obviously due to high losses of *p*-HAPG acids eluted earlier than *p*-HAP- β -D-glucopyranoside, *p*-HAP earlier than trifluoroacetyl-*p*-HAP and trifluoroacetyl-*p*-HAP earlier than acetyl-*p*-HAP (Table I).

The precision and accuracy of the measurement by HPLC as shown in Table II demonstrate the high reliability of the HPLC method in the analysis of *p*-HAP and *p*-HAPG in needle extracts. Table III shows the comparison of the measurement of *p*-HAP and *p*-HAPG by HPLC- and TLC-GC methods. Lower values of *p*-HAPG were measured by the TLC-GC method, obviously due to high losses of *p*-HAPG in different steps of sample preparation for GC measurement (extraction, fractionation, TLC, enzymic hydrolysis, extraction)².

GC-ECD of trifluoroacetylated extracts before and after alkaline hydrolysis following purification by TLC confirmed the presence of *p*-HAP and *p*-HAP- β -D-

TABLE III

COMPARISON BETWEEN HPLC AND GC METHODS IN THE MEASUREMENT OF *p*-HAP AND *p*-HAPG IN NEEDLES3-Year-old needles, 100-year old healthy trees, FA Sauerlach, harvest: June 12th, 1984, *p*-HAP: $\mu\text{g/g}$ fresh weight, *p*-HAPG: $\mu\text{g/g}$ fresh weight *p*-HAP equivalents.

Sample no.	GC-FID		HPLC-UV	
	<i>p</i> -HAP	<i>p</i> -HAPG	<i>p</i> -HAP	<i>p</i> -HAPG
1	56	405	63	405
2	192	1750	157	2763
3	137	568	131	912
4	29	1194	18	2546
5	251	409	352	566
6	41	1466	20	1840
7	134	164	105	229

glucopyranoside (trifluoroacetyl-*p*-HAP, GC-ECD, $t_R = 7.2$ min) in different tissues of needles as shown by HPLC.

The response of the electron-capture detector was found to be linear in the measuring range 5–62 ng and can be described by the regression line, $y = -22.07 + 0.16x$ ($r^2 = 0.82$), where y = injected amount of trifluoroacetyl-*p*-HAP (ng) and x = detected area of the peak (mm^2). However, the GC-ECD technique was found to be unsuitable for a large number of samples.

Previously, the identification of *p*-HAP was reported by us in Norway spruce needles³. *p*-HAPG was also unequivocally identified⁹ in Norway spruce needles using UV, IR, mass spectroscopy ¹H NMR and ¹³C NMR spectroscopy. Table IV shows the levels of *p*-HAP and *p*-HAP- β -D-glucopyranoside in different tissues and needles as measured by HPLC.

TABLE IV

CONTENTS OF *p*-HAP AND *p*-HAPG IN EPIDERMAL STRIPS ($\mu\text{g/g}$ FRESH WEIGHT), VASCULAR BUNDLES ($\mu\text{g/g}$ DRY WEIGHT) AND NEEDLES OF CURRENT-YEAR SHOOTS ($\mu\text{g/g}$ FRESH WEIGHT) BY HPLC

Needles/parts of needles	Tree No.	Plant No.*	<i>p</i> -HAP	<i>p</i> -HAPG (<i>p</i> -HAP equivalents)
Epidermal strips	1		353	545
	2		744	6185
	3		745	1346
Vascular bundles	1		116	250
	2		—	74
Needles of current-year shoots		1	274	8
		2	322	35
		3	372	0
		4	552	58

* Plants with planting shock.

CONCLUSIONS

The results show that the determination of *p*-HAP and *p*-HAP- β -D-glucopyranoside can be conveniently carried out by isocratic HPLC. This method is superior to the methods described previously¹⁻⁴. The following merits of the isocratic method are important: rapid measurement (8-10 samples/day), no enzymic hydrolysis of *p*-HAP- β -D-glucopyranoside necessary prior to HPLC measurements, sensitive measurement in the nanogramme range, low cost/measurement, measurement of *p*-HAP and *p*-HAPG in the same run.

A further reduction of analysis duration could be obtained with 2,5-dihydroxyacetophenone (retention time 27.2 min) instead of acetyl-*p*-hydroxyacetophenone (retention time 55.0 min, see Table I) as internal standard. The response of the detector was found to be linear for 2,5-dihydroxyacetophenone in the range 20-840 ng.

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